

Stereoselective Synthesis of Spiro[tetrahydropyran-3,3′-oxindole] Derivatives Employing Prins Cascade Strategy

B. V. Subba Reddy,^{*,†} V. Swathi,^{†,‡} Manisha Swain,[†] Manika Pal Bhadra,[‡] B. Sridhar,[§] D. Satyanarayana,[⊥] and B. Jagadeesh $¹$ </sup>

†Natural Product Chemistry, [‡]Centre for Chemical Biology, [§]Laboratory of Crystallography, and [⊥]Centre for Nuclear Magnetic Resonance and Structural Chemistry, CSIR-Indian Institute of Chemical Technology, Tarnaka, Hyderabad 500 607, India

S Supporting Information

[AB](#page-2-0)STRACT: [A variety o](#page-2-0)f aldehydes undergo smooth coupling with 4-hydroxy-N-methyl-2-methylene-N-phenylbutanamide in the presence of BF_3 ·OEt₂ under ambient conditions to produce the corresponding spiro-oxindole derivatives in good yields with excellent selectivity. It is an entirely new strategy to construct the spirocycles in a one-pot operation through a Prins cascade process.

The spiro-oxindoles are important synthetic targets that are
often found in various biologically active molecules

Figure 1. Examples of biologically active spiro-oxindoles.

^aYield refers to pure product after column chromatography. ^bRatio was determined by ¹ H NMR.

Figure 2. Schematic structure of major product 6m (a) and the observed characteristic NOEs (b).

Figure 3. Overlay of the lowest energy NMR structures in solution state (a) and X-ray crystal structure (b) of major product 6m.

(Figure 1).¹ They are considered to be privileged scaffolds in medicinal chemistry due to their broad range of biological activities i[n](#page-2-0)cluding anticancer, antimalarial, antituberculosis, growth hormone secretagogue, and progesterone receptor agonists.² Consequently, several approaches have been developed for the stereoselective synthesis of spiro-oxindole derivativ[es](#page-2-0), which are being used as broad and promising synthetic scaffolds in several therapeutic areas.^{3−5} However, a few methods are reported for the synthesis of tetrahydrospiro- [indoline-3,4′-pyran]-2-one derivatives.^{6−9}

Received: October 22, 2014 Published: December 8, 2014

ACS Publications

Figure 4. ORTEP diagram of 6m.

Table 2. Preparation of Spiro-oxindoles^a

Following our interest in spirocycles, 10 we herein report a novel strategy for the stereoselective synthesis of tetrahydrospiro[indoline-3,4′-pyran]-2-o[ne](#page-2-0) derivatives through a Prins cascade cyclization.¹¹ Initially, we attempted the coupling of 4-hydroxy-N-methyl-2-methylene-N-phenylbutanamide (4) with p-chlorobenzal[deh](#page-2-0)yde (5) in the presence of 0.3 equiv of In(OTf)₃ in dichloromethane at 0 °C (Table 1, entry a). However, no cyclization was observed under these conditions. A similar result was also observed with 0[.3](#page-0-0) equiv of $Sc(OTf)$ ₃ (Table 1, entry c). Even 1.2 equiv of $In(OTf)$ ₃ was found to be insufficient in promoting the cyclization reaction (Table 1, entry b). [N](#page-0-0)ext, we performed the reaction using 1.2 equiv of TMSOTf at room temperature. Interestingly, the desired [p](#page-0-0)roduct was obtained in 90% yield as a mixture of

 a Products were characterized by NMR, IR, and mass spectrometry. b Yield refers to the pure product after column chromatography. b Ratio of the product was determined by $^1\mathrm{H}$ NMR.

Scheme 1. Plausible Reaction Pathway

diastereomers 6a and 7a in 85:15 ratio (Table 1, entry d). The diastereomeric ratio was increased to 9:1 when the reaction was performed at 0 °C to room temperature (T[ab](#page-0-0)le 1, entry e). Finally, the best results (92% yield and 96:04 ratio) were obtained with 1.2 equiv of BF_3 ·OEt₂ at 0 °[C](#page-0-0) to room temperature (Table 1, entry g). Surprisingly, no cyclization was observed at −20 °C (Table 1, entry h). Further, we screened the solvents for thi[s](#page-0-0) spirocyclization. Among various solvents such as dichloromethane, [ac](#page-0-0)etonitrile, tetrahydrofuran, and toluene, DCM gave the best results in terms of conversion.

The solution state structure of the major product 6m was derived by using detailed NMR studies such as 2D-NOESY, COSY, TOCSY, and J-coupling constants. The measured strong scalar coupling values, $\frac{3}{1-H/2}$ _H = 11.8 and $\frac{3}{4}$ _{H/5-H} = 13.8 Hz, from the one-dimensional ¹H NMR spectrum clearly indicate that the 1-H, 2-H, 4-H, and 5-H protons are in axial positions in the six-membered ring. The observed characteristic NOE correlations, 1-H/5-H, 2-H/4-H, 2-H/OCH₃(a) and 4- $H/OCH₃(a)$, and the scalar coupling values confirm that the six-membered ring is in the chair conformation and exhibits the relative stereochemistry as shown in Figure 2.

The lowest energy structures of major product 6m, derived from the NOE-restrained MD simulations, [ar](#page-0-0)e compared with X-ray crystal structure and are found to be in excellent agreement with the solid state structure (Figures 3 and 4).

Inspired by the above results, we extended this method for different substrates bearing various substituents on [th](#page-0-0)e ar[om](#page-1-0)atic ring of the aldehyde. A variety of aldehydes including aromatic, heteroaromatic, and aliphatic aldehydes underwent smooth cyclization under the optimized conditions (Table 2).

A plausible mechanism for the Prins cascade process is proposed in Scheme 1. We assume that the reacti[on](#page-1-0) proceeds through an oxo-carbenium ion that is formed in situ from alkenol and aldehyde after likely activation with BF_3 ·OEt₂. A subsequent attack of the olefin on the oxo-carbenium ion followed by the trapping of carbocation with a tethered aryl group would give the desired spirocycle.

Generally, the Prins cyclization proceeds with electron-rich olefins, whereas, in the present study, we showed that it also proceeds with electron-deficient olefins. This may be attributed to the formation of oxindole with simultaneous Prins-type cyclization.

In summary, we have developed a novel strategy for the stereoselective construction of spiro-oxindoles from 4-hydroxyN-methyl-2-methylene-N-phenylbutanamide and aldehydes through a Prins cascade cyclization. This method offers significant advantages such as mild reaction conditions, broad substrate scope, high conversions, and excellent diastereoselectivity, which make it quite simple, more convenient, and practical. It provides the spirocycles that are of increasing interest to the pharmaceutical industry.

■ ASSOCIATED CONTENT

S Supporting Information

Copies of ¹H and ¹³C NMR spectra of products (6a–6p), NOESY, and DQFCOSY study of compounds (6m). This material is available free of charge via the Internet at http:// pubs.acs.org.

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: basireddy@iict.res.in.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

V.S. thanks UGC, and M.S. thanks the Council of Scientific and Industrial Research (CSIR), New Delhi, for the award of fellowships. B.V.S. thanks CSIR, New Delhi, for the financial support as a part of XII five year plan program under title ORIGIN (CSC-0108).

■ REFERENCES

(1) Badillo, J. J.; Hanhan, N. V.; Franz, A. K. Curr. Opin. Drug Discovery Dev. 2010, 13, 758.

(2) (a) Singh, G. S.; Desta, Z. Y. Chem. Rev. 2012, 112, 6104. (b) Dalpozzo, R.; Bartoli, G.; Bencivenni, G. Chem. Soc. Rev. 2012, 41, 7247.

(3) (a) O'Brien, J. M.; Kingsbury, J. S. J. Org. Chem. 2011, 76, 1662. (b) Franz, A. K.; Hanhan, N. V.; Ball-Jones, N. R. ACS Catal. 2013, 3, 540.

(4) Trost, B. M.; Brennan, M. K. Synthesis 2009, 3003.

(5) (a) Antonchick, A. P.; Gerding-Reimers, C.; Catarinella, M.; Schürmann, M.; Preut, H.; Ziegler, S.; Rauh, D.; Waldmann, H. Nat. Chem. 2010, 735. (b) Ashimori, A.; Overman, L. E. J. Org. Chem. 1992, 57, 4571.

(6) Wang, J.; Crane, E. A.; Scheidt, K. A. Org. Lett. 2011, 13, 3086. (7) (a) Castaldi, M. P.; Troast, D. M.; Porco, J. A. Org. Lett. 2009, 11,

3362. (b) Zhang, Y.; Panek, J. S. Org. Lett. 2009, 11, 3366.

(8) Shintani, R.; Hayashi, S.-Y.; Murakami, M.; Takeda, M.; Hayashi, T. Org. Lett. 2009, 11, 3754.

(9) (a) Hsieh, J.-C.; Cheng, A.-Y.; Fu, J.-H.; Kang, T.-W. Org. Biomol. Chem. 2012, 10, 6404. (b) Wang, H.; Guo, L.-N.; Duan, X.-H. Org. Lett. 2013, 15, 5254.

(10) (a) Reddy, B. V. S.; Karthik, G.; Rajasekaran, T.; Antony, A.; Sridhar, B. Tetrahedron Lett. 2012, 53, 2396. (b) Reddy, B. V. S.; Rajasekaran, T.; Karthik, G.; Rao, T. P. Tetrahedron Lett. 2012, 53, 3416. (c) Rajasekaran, T.; Karthik, G.; Sridhar, B.; Reddy, B. V. S. Org. Lett. 2013, 15, 1512. (d) Reddy, B. V. S.; Kumar, H.; Reddy, P. S. K.; Kumar, S. K. Eur. J. Org. Chem. 2014, 4234. (e) Reddy, B. V. S.; Durgaprasad, M.; Sridhar, B.; Kumar, S. K. J. Org. Chem. 2014, 79, 2289.

(11) Overman, L. E. Aldrichimica Acta 1995, 28, 107.

■ NOTE ADDED AFTER ASAP PUBLICATION

Scheme 1 has been updated. The revised version was re-posted on December 19, 2014.